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Tetrahedron Letters 45 (2004) 7301-7302

Tetrahedron Letters

## 1,3-Dipolar addition of diethyl diazomethylphosphonate onto a C=N double bond. Synthesis of triazolinyl and aziridinyl phosphonates

Romuald Bartnik, Stanisław Leśniak\* and Piotr Wasiak

Department of Organic and Applied Chemistry, University of Łódź, Narutowicza 68, 90-136 Łódź, Poland

Received 2 June 2004; revised 28 July 2004; accepted 3 August 2004

Abstract—Cycloaddition of diethyl 1-diazomethylphosphonate onto a C=N double bond is described. It was found that the reaction of this phosphorus diazomethane with benzylidene-alkyl-amines resulted in the formation of (1-alkyl-5-phenyl-4,5-dihydro-1H-[1,2,3]triazol-4-yl)-phosphonic acid diethyl esters, whereas its reaction with methylene-aryl-amines, generated from the corresponding hexahydro-1,3,5-triazines, led to diethyl esters of (aziridin-2-yl)-phosphonic acid. © 2004 Elsevier Ltd. All rights reserved.

Aziridine-2-carboxylates are important building blocks for the synthesis of  $\alpha$ - or  $\beta$ -amino acids as they can undergo highly regiocontrolled and stereoselective ring-opening reactions.<sup>1</sup> Aziridine-2-phosphonates are expected to play a similar role in the synthesis of  $\alpha$ - or β-amino phosphonates.<sup>1</sup> Although methods for the synthesis of aziridine-2-carboxylate esters are well established,<sup>2</sup> there are only a few reports on the synthesis of their P-counterparts. The first was obtained in moderate yield from diethylvinylphosphonate.<sup>3</sup> Racemic aziridine 2-phosphonates have been prepared by the copper catalysed aziridination of vinylphosphonates with [N-(ptoluenesulfonyl)imino]-phenyliodonane4 or N-[(4-nitrobenzene)sulfonyl]oxy-carbamate<sup>5</sup> and the Darzens-type addition of lithiated chloromethylphosphonate anions to an imine<sup>6</sup> or a sulfinimine.<sup>7</sup> Moreover, 1-vinyl-2phosphono-aziridines were obtained by the reaction of diazomethane with 1-phosphono-2-aza-1,3-dienes.8

Phosphorus diazomethanes have been the most frequently used phosphorus reagents for efficient onecarbon homologation of aldehydes/ketones to terminal/ internal alkynes but, to the best of our knowledge, the reaction of these reagents with a C=N double bond has never been applied in the synthesis of three- or five-membered rings.

In this letter we describe our preliminary results on the synthesis of  $\Delta^2$ -1,2,3-triazolinyl-4- and aziridinyl-2-phosphonates by the cycloaddition of diethyl diazomethyl-phosphonate<sup>9</sup> on to a C=N double bond.

Our initial work focused on the reaction of diethyl diazomethylphosphonate 1 with benzylidene-*N*-alkyl-amines **2a–d**. The reaction with **2a** was carried out in various solvents at room temperature over several days and monitored by <sup>1</sup>H NMR spectroscopy. The best result was obtained in methanol for three days in darkness (Scheme 1), giving (1-methyl-5-phenyl-4,5-dihydro-1*H*-[1,2,3]triazol-4-yl)-phosphonic acid diethyl ester **3a** in 77% yield<sup>10</sup> as a single isomer. Similarly, products **3b–d** were obtained in 70–80% yields (Table 1). The configuration of the ring substituents was not studied in detail but the high value of coupling constants *J* (11.7Hz) between protons at C-4 and C-5 suggests their *trans* arrangement.<sup>11</sup>



Scheme 1.

Keywords: Triazolines; Aziridines; Phosphonic diethyl esters; Cycloaddition.

<sup>\*</sup> Corresponding author. Tel.: +48 42 635 57 65; fax: +48 42 678 16 09; e-mail: slesniak@chemul.uni.lodz.pl

<sup>0040-4039/\$ -</sup> see front matter @ 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2004.08.016

Table 1. Preparation of the 4,5-dihydro-1H-[1,2,3]triazol-4-yl)-phosphonic acid diethyl esters **3a**-d from imines **2a**-d

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	Entry	Imine	R	Product (yield, %) <sup>a</sup>
	1	2a	Me	<b>3a</b> (78)
	2	2b	Et	<b>3b</b> (71)
	3	2c	<i>i</i> -Pr	<b>3c</b> (82)
	4	2d	t-Bu	<b>3d</b> (83)

<sup>a</sup> All the yields refer to isolated products.

Products **3a–d** were purified chromatographically and characterised by spectroscopic methods. It should be noted that only one method of the synthesis of 1,2,3-triazoline systems containing a diethyl or dimethyl phosphonate had previously been described by reactions of aryl azides with vinylphosphonates.<sup>11</sup> On the other hand our method permits preparation of 1-alkyl- but not 1-aryl-triazolines. All attempts of reactions in question using *N*-aryl imines were unsuccessful.

Although a thermal ring contraction of 1,2,3-triazolines to aziridines is a well-established process,<sup>12</sup> all attempts of N<sub>2</sub>-elimination from triazines 3a-d by thermolysis or photolysis failed. Thus, thermolyses performed under FVT conditions or in high boiling solvents afforded only imines 2 as retro-addition products. In all cases, UVirradiation of 3 gave complicated mixtures of many components, however, the formation of desired aziridines was not observed.

In parallel investigations, we also studied the cycloaddition reactions of **1** with hexahydro-1,3,5-triazines as precursors of reactive *N*-methylenamines. When mixtures of **1** and 0.33 equiv of 1,3,5-triaryl-hexahydro-1,3,5-triazine: **4a**–**d** in methanol were left for 15 days in darkness (the progress of reactions was monitored by <sup>1</sup>H NMR), the corresponding aziridines **5a**–**d** were obtained in good yields (Scheme 2, Table 2).

These products were purified by chromatography and identified by spectroscopic methods.<sup>13</sup>

The results show that the cycloaddition of 1 onto a C=N double bond formed in situ, affords the corre-



Scheme 2.

Table 2. Preparation of the (1-aryl-aziridin-2-yl)-phosphonic acid diethyl esters 5a-d

Entry	Triazine	Ar	Product (% yield) <sup>a</sup>
1	<b>4</b> a	$C_6H_5$	<b>5a</b> (67)
2	4b	p-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	<b>5b</b> (70)
3	4c	p-ClC <sub>6</sub> H <sub>4</sub>	<b>3c</b> (78)
4	4d	o-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	<b>3d</b> (74)

<sup>a</sup> All the yields refer to isolated products.

sponding triazolines, which undergo a spontaneous extrusion of  $N_2$  to aziridines 5. The fact that these reactions occurred in the absence of a catalyst is remarkable.

In summary, addition of diethyl 1-diazomethylphosphonate onto C=N double bonds seems to provide a practical alternative to the existing methods to prepare azacycles with a phosphonate group. Further studies directed towards an application of various catalysts in this reaction are in progress.

## Acknowledgements

This research project was supported by the grant no. 505/681/2003 from University of Łódź.

## **References and notes**

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- 10. For example **3a**. Purified by chromatography on Al<sub>2</sub>O<sub>3</sub> (hexane/AcOEt). Colourless oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>): 1.29 (t, *J* 7Hz, 3H), 1.34 (t, *J* 7Hz, 3H), 3.20 (s, 3H), 4.13–4.27 (m, 4H), 4.54 (dd, <sup>3</sup>*J*<sub>HH</sub> 11.7Hz, <sup>3</sup>*J*<sub>HP</sub> 19.6Hz, 1H), 4.55 (dd, <sup>3</sup>*J*<sub>HH</sub> 11.7Hz, <sup>2</sup>*J*<sub>HP</sub> 30.1Hz, 1H), 7.24–7.39 (m, 5H). <sup>13</sup> C NMR (CDCl<sub>3</sub>): 16.40 (d, <sup>3</sup>*J*<sub>CP</sub> 4.6Hz, CH<sub>2</sub>CH<sub>3</sub>), 16.48 (d, <sup>3</sup>*J*<sub>CP</sub> 4.3Hz, CH<sub>2</sub>CH<sub>3</sub>), 35.17, (NCH<sub>3</sub>), 62.78 (d, <sup>2</sup>*J*<sub>CP</sub> 6.9Hz, CH<sub>3</sub>CH<sub>2</sub>), 63.45 (d, <sup>2</sup>*J*<sub>CP</sub> 6.9Hz, CH<sub>3</sub>CH<sub>2</sub>), 64.95 (*C*<sub>5</sub>), 82.73 (d, <sup>1</sup>*J*<sub>CP</sub> 161.2Hz, *C*<sub>4</sub>), 126.90, 128.46, 128.93 (3 × *C*<sub>ar</sub>), 138.11 (d, <sup>3</sup>*J*<sub>CP</sub> 10.3Hz, *C*<sub>9</sub>). <sup>31</sup> P NMR (CDCl<sub>3</sub>): 19.32. [Found: C, 52.74; H, 6.99; N, 14.00. C<sub>13</sub>H<sub>20</sub>N<sub>3</sub>O<sub>3</sub>P (289.37) requires C, 52.52; H, 6.78; N, 14.13%].
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- 13. For example **5a**. Purified by SiO<sub>2</sub>-chromatography (hexane/AcOEt). Colourless oil. <sup>1</sup> H NMR (CDCl<sub>3</sub>): 1.33 (t, J 6.9 Hz, 6H), 2.03–2.70 (m, 3H), 4.10–4.43 (m, 4H), 6.85–7.40 (m, 5H). <sup>13</sup> C NMR (CDCl<sub>3</sub>): 16.45 (CH<sub>3</sub>CH<sub>2</sub>), 16.69 (CH<sub>3</sub>CH<sub>2</sub>), 31.65 (<sup>2</sup>J<sub>CP</sub> 4.9 Hz, C<sub>3</sub>), 32.68 (<sup>1</sup>J<sub>CP</sub> 217.3 Hz, C<sub>2</sub>), 62.81 (d, <sup>2</sup>J<sub>CP</sub> 6.1 Hz, CH<sub>3</sub>CH<sub>2</sub>), 63.12 (d, <sup>2</sup>J<sub>CP</sub> 6.1 Hz, CH<sub>3</sub>CH<sub>2</sub>), 120.91, 123.56, 129.36 (3 × C<sub>ar</sub>), 153.65 (d, <sup>3</sup>J<sub>CP</sub> 6.1 Hz, C<sub>q</sub>). IR (neat): 2984, 1599, 1492, 1453, 1392, 1368, 1358, 1256, 1165, 1097, 1052, 1026, 966, 797, 766, 730, 696. MS (*m*/e, %): 255 (M<sup>+</sup>, 87), 226, 198, 182, 168, 147, 146, 132, 119, 118, 117, 109, 105, 104, 92, 91(100), 81, 77, 65, 51. HRMS: calcd for C<sub>12</sub>H<sub>18</sub>NO<sub>3</sub>P M 255.10192, found M<sup>+</sup>· 255.10243.